# CHEMICAL COMPOSITION, ANTIOXIDANT AND ANTIMICROBIAL ACTIVITY OF THE TURMERIC ESSENTIAL OIL (Curcuma longa L.)

Jelena S. Stanojević, Ljiljana P. Stanojević\*, Dragan J. Cvetković, Bojana R. Danilović

Faculty of Technology, Leskovac, Serbia

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In the present work, essential oil has been obtained by Clevenger-type hydrodestillation from grounded curcuma rhizome (Rhizoma Curcumae) (Turkey) with hydromodulus 1:5 m/V during 180 minutes. The qualitative and quantitative composition of the oil was determined by GC-MS and GC-FID spectrometry. The antioxidant activity of the obtained oil was determined using DPPH assay just after adding DPPH radical and after 20 min, 30 min and 45 min incubation with radical. The antimicrobial activity was determined using a discdiffusion method. The yield of the essential oil was 0.3 cm<sup>3</sup>/ 100 g plant material. Eight compounds were identified. The major ones were ar-turmerone (22.7%), turmerone (26%) and curlone (16.8%). The best antioxidant activity showed the oil incubated for 45 minutes with DPPH radical. EC50 values for the obtained oil were 1.784 mg/cm3 (without incubation), 0.098 mg/cm3 (after 20 minutes), 0.072 mg/cm³ (after 30 minutes) and 0.045 mg/cm³ (after 45 minutes incubation with radical). The oil showed the best antimicrobial activity against Candida albicans. The results indicate that turmeric essential oil is an extremely strong antioxidant and antimicrobial (antifungal) agent with potential application in the food and pharmaceutical industries as a safer alternative to the synthetic antioxidants and antimicrobial agents.

**Keywords:** Curcuma longa L., Clevenger hydrodistillation, Antioxidant activity, Antimicrobial activity, GC-MS

#### Introduction

Essential oils are complex mixtures of secondary plant metabolites. They are highly concentrated, volatile, oily distillates. Essential oils are found in all plant parts (flowers, barks, roots, leaves, peels, seeds) due to the activity of endogenous and exogenous secretory plant tissues. Among many other methods, essential oils could be obtained by hydrodistillation, steam, water/steam distillation, expression and extraction with supercritical carbon dioxide [1,2]. Owing to their aroma, odor and a plethora of beneficial effects, they are widely used in perfumes, cosmetics, aromatherapy and nutrition [3,4].

Curcuma longa L., also known as curcuma (turmeric, indian shaffron, golden Goddess) is a perennial herbaceous aromatic plant from the ginger family (Zingiberaceae). It is assumed that turmeric originated in China, and Buddhist monks or Chinese migration brought it to the Indian subcontinent. Anyway, turmeric is nowadays cultivated in Asian countries (Bangladesh, China, Thailand, Cambodia, Malaysia, Indonesia, Philippines) and some parts of South America (Peru and Bolivia) but India still remains the largest producer, consumer and exporter [5]. The plant has yellow flowers and reaches a height of about 1 m. The underground rhizome is yellowish, consisting of two main parts: the egg-shaped (mother) rhizome and the long cylindrical, branched primary, secondary and even tertiary rhizomes

[6,7]. Turmeric rhizome contains two major classes of secondary metabolites: phenolic curcuminoids and essential oil [8]. These metabolites are largely responsible for the pharmacological effects of turmeric [9]. The composition of the both metabolites depends on a genotype, the environment, harvest season, dry process and storage conditions [10]. Curcuminoids are responsible for the yellow color of the turmeric, and the essential oil that it contains for its aroma and taste [7,11]. The major and the most studied curcuminoid found in turmeric is curcumin, which is recognized as the most responsible compound for the majority of beneficial effects which this miraculous plant exhibits. Besides curcumin, there are two more curcuminoids: demetoxycurcumin and bisdemetoxycurcumin [12]. The essential oil could be obtained from fresh [9,13,14] and dry leaves [15,16], fresh flowers [16], dry roots [16] and fresh [9,13,17] and dry rhizomes [16.18] of turmeric. Dried rhizomes and leaves are used for the essential oil extraction in the industry. Rhizomes (despite the fact that they contain a higher amount of active compounds in comparison to other plant parts) [13] have a higher oil content than leaves, 5-6% vs. 1-1.5%, respectively [7]. The essential oils from leaves and flowers are dominated by monoterpenes while those from roots and rhizomes primarily contained sesquiterpenes

\*Author address: Ljiljana Stanojević, Faculty of Technology, Bulevar oslobodjenja 124, 16000 Leskovac, Serbia

E-mail: Ijiljas76@yahoo.com

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[19]. The major volatile principles of the rhizome oil are  $\alpha$ - and  $\beta$ -turmerone and ar-turmerone [10].

Turmeric use dates from even 6000 years ago. It exhibits many beneficial effects because of phytochemicals that it contains and some of them are: anti-carcinogenic, anti-inflammatory, anti-microbial [20], anti-fungal [21], anti-mutagenic [22], hypocholesteremic, insect repellent, anti-rheumatic, anti-fibrotic, anti-venomous, anti-diabetic, anti-viral, anti-hepatotoxic [5]. Turmeric was (and is) used for religious proposes (as an amulet) in the Hindu culture; as spice and food colorant (because of its flavor and golden color) and a food preservative in India; in Ayurveda (traditional Indian medicine) it is used orally as stomachic and blood purifier, to treat gall bladder and heart problems, liver disorders, bloating, menstrual problems, disorders of the urinary tract, allergies, arthritis and other chronic diseases [11], topically in the treatment of various skin diseases [11,19] and via inhalation in treating chronic rhinitis and coryza [19].

The main objectives of this study were to determine the chemical composition of the essential oil obtained from the turmeric rhizome by Clevenger hydrodistillation using GC-MS method, as well as its antioxidant activity using DPPH assay, and antimicrobial activity using a disc-diffusion method in order to contribute to the greater use of turmeric in pharmaceutical and food industry in the southeast Serbia. Although many researchers worked on the isolation and characterization of turmeric essential oil, according to the literature cited in this paper, the determination of the chemical composition and investigation of the biological activity of the essential oil isolated from the commercial spice originated from Turkey has not been done yet. This is reasonable to some extent, having in mind the fact that Turkey is not in the top list of the turmeric producer and consumer countries.

# Experimental -

### Plant material

A fine, yellow turmeric powder (commercial spice) obtained from the dry, ground turmeric rhizome (*Rhizoma Curcumae*) was purchased from the local health food store in Vlasotince, Serbia. According to the declaration it originates from Turkey (importer: Balkan Komerc; packed by: BIOTIKA d.o.o. Beograd, Bulevar Vojvode Mišića 37-39 in collaboration with RADAKOVIĆ CO d.o.o. Vršac).

#### Reagents and chemicals

Ethanol, 96% (Centrohem, Zemun, Serbia), 1.1-diphenyl-2-picrylhydrasil (DPPH radical), butylated hydroxytoluene (BHT) (Sigma Chemical Company, St. Louis, USA), dimethyl sulfoxide (DMSO; BDH, Milan, Italy).

#### Isolation of the essential oil

Essential oil was obtained by Clevenger hydrodistillation, with hydromodulus 1:5 m/V during 180 minutes. The amount of the essential oil was determined per 100 g of

the plant material. The obtained oil was dried over anhydrous sodium sulfate and kept at 4 °C until analysis.

## GC-MS and GC-FID analysis

GC-MS analysis of the essential oil obtained from turmeric rhizome was performed on Agilent Technologies 7890B gas chromatograph, equipped with weakly polar, silica capillary column, HP-5MS (5% diphenyl- and 95% dimethyl-polysiloxane, 30 m x 0.25 mm, 0.25 µm film thickness; Agilent Technologies, USA) and coupled with inert, selective 5977A mass detector of the same company. One µl of the sample dissolved in diethyl ether in the concentration of 500 ppm was injected in 20:1 split mode. Helium was used as the carrier gas, at a constant flow rate of 1 cm<sup>3</sup>/min. The oven temperature was programmed from 50 °C for 2.25 minutes and then increased to 290 °C at the rate of 4 °C/min. Temperatures of the MSD transfer line, ion source and quadruple mass analyzer were set at 300 °C, 230 °C and 150 °C, respectively. The ionization voltage was 70 eV and mass range m/z 35-650.

GC-FID analysis was carried out under identical experimental conditions as GC-MS. The temperature of the flame-ionization detector (FID) was set at 300 °C.

Data processing was performed using MSD Chem-Station, MassHunter Qualitative Analysis and AMDIS 32 softwares (Agilent Technologies, USA). Retention indices of the components from the analyzed samples were experimentally determined using a homologous series of n-alkanes from C8-C20 as standards. Compounds identification was based on the comparison of their retention indices (RI<sup>exp</sup>-Table I) with those available in literature [23,24] (RI<sup>iit</sup>-Table I), as well as their mass spectra with those from Willey, NIST and RTLPEST libraries. The percentage composition of particular components in the essential oil was determined on the basis of automatically integrated peak areas of the GC-FID signal.

# DPPH assay

The ability of the essential oil to scavenge free DPPH radicals was determined using the DPPH assay. The essential oil was dissolved in ethanol and a series of different concentrations was prepared. The ethanol solution of DPPH radical (1 cm³, 300 µmol solution (3x10-4 mol/dm³)) was added to 2.5 cm³ of the prepared essential oil solutions. The absorption was measured at 517 nm immediately after adding the DPPH radical and after 20, 30 and 45 minutes incubation with radical. The absorption at 517 nm was determined for the ethanolic solution of DPPH radical as well, which was diluted in the aforementioned ratio (1 cm³ of the DPPH radical of the given concentration with 2.5 cm³ ethanol added). Ethanol was used as a blank. Free radical scavenging activity was calculated according to the formula:

DPPH radical scavenging capacity (%) = 100 - 
$$\left[ \left( A_{\rm S} - A_{\rm B} \right) \times \frac{100}{A_{\rm C}} \right]$$

As – Absorption of the "sample" at 517 nm. "Sample" – ethanolic solution of the essential oil treated with DPPH radical solution

AB – Absorption of the "blank" at 517 nm. "Blank" – ethanolic solution of the essential oil which is not treated with DPPH radical solution

Ac – Absorption of the "control" at 517 nm. "Control" – ethanolic solution of the DPPH radical

All absorptions were measured on UV-VIS VARIAN-Cary 100 Conc. Spectrophotometer.

The essential oil concentration needed for the neutralization of 50% of the initial DPPH radical concentration is called EC50 value. This value was determined by using a linear regression analysis in the concentration range between 0.008 and 2 mg/cm<sup>3</sup> of the essential oil added to the reaction mixture.

## Antimicrobial activity

Microorganisms and mediums

Eight microorganisms were selected to determine the antimicrobial activity of turmeric essential oil: Candida albicans (ATCC 10259), Proteus vulgaris (ATCC 8427), Bacillus cereus (ATCC), Bacillus subtilis (ATCC 6633), Klebsiella pneumoniae (ATCC 700603), Staphylococcus aureus (ATCC 25923), Escherichia coli (ATCC 25922) and Listeria monocytogenes (ATCC 19166). Mediums used for the growth of the microorganisms: nutrient agar for bacterial growth and Sabouraud maltose agar (Torlak, Belgrade) for fungi. Microorganisms are from the collection of the Microbiology Laboratory, Faculty of Technology, Leskovac.

### Disc-diffusion method

The agar disc-diffusion method was used for testing the antimicrobial activity of turmeric essential oil [25]. The mediums were sterilized for 15 minutes in an autoclave at 121 °C under 110 kPa.

An inoculum of 0.1 cm³ of overnight culture was added to 10 cm³ of the medium and poured into petri dishes. For screening, sterilised filter paper disks (12.7 mm dia., Schleicher&Schuell) were placed on the surface of inoculated mediums and impregnated with 60 µl of the essential oil (1:10 V/V in DMSO). The plates were incubated for 24 hours at 37 °C for bacteria, and 48 hours at 25 °C for fungi. After incubation, the inhibition zone diameters were measured and expressed in mm. The presence of the inhibition zone indicates the activity of the tested samples against bacteria or fungi.

Standardized discs of Ampicilin (10 µg/disc), Bactrim (25 µg/disc), Cefalexin (30 µg/disc) (Bio Rad) and Nystatin (100 U/disc) (Bioanalyse) were used as reference standards. DMSO was used as negative control.

#### **Results and Discussion**

Qualitative and quantitative essential oil composition The essential oil from the ground turmeric rhizome (turmeric spice) was isolated by Clevenger hydrodistillation. The yield of the essential oil was  $0.3~\text{cm}^3$  per 100~g plant material. GC-MS analysis resulted in identifying eight compounds: eugenol, (*E*)-cariophyllen, curcumen,  $\alpha$ -zingiberene,  $\beta$ -sesquiphelandrene, ar-turmerone, turmerone and curlone, representing 82.9% of the total oil composition.

Table 1. Chemical composition of the essential oil from turmeric rhizome

Peak	t <sub>ret.</sub> , min	Component	Rlexp	Rl <sup>lit</sup>	Composition (%)
		Benzene derivatives (8%)			
1	24.8766	Eugenol	1360	1356a	8
		Sesquiterpene Hydrocarbons (9.5%)			
2	26.9897	( <i>E</i> )-Caryophyllene	1427	1417a	2
3	28.8137	ar-Curcumene	1487	1479a	1.8
4	29.1743	α-Zingiberene	1499	1493ª	2.8
5	30.0380	β-Sesquiphellandrene	1529	1521a	2.8
		Oxygenated sesquiterpenes (65.4%)			
6	34.0390	ar-turmerone	1671	1668ª	22.7
7	34.1516	Turmerone	1675	1674 <sup>b</sup>	26
8	35.0389	Curlone	1708	1706 <sup>b</sup>	16.8
				Total	82.9%

t<sub>ret.</sub>: Retention time; RI<sup>lit</sup> a,b-Retention indices from literature [23,24], respectively; RI<sup>exp</sup>: Experimentally determined retention indices using a homologous series of n-alkanes (C8-C20) on the HP-5MS column.

All identified compounds were designated by numbers from 1-8 according to their elution order and classified into benzene derivatives, sesquiterpene (C15) hydrocarbons and oxygenated sesquiterpens together with their percentage compositon given in Table 1. The structures of the identified components are given in Figure 1.

The major compounds were oxygenated sesquiterpenes comprising up to 65.4% represented by turmerone, ar-turmerone and curlone (Table 1), followed by benzene derivatives with 8% and sesquiterpene hydrocar-

bons with 9.5% of the total oil composition. Leela and coworkers [16] obtained simular results. Namely, they used fresh flowers and dried leaves, root and rhizome from  $Curcuma\ longa$  for obtaining essential oils by Clevenger hydrodestilation. The major components in the essential oil from dried rhizome were ar-turmerone, turmerone, curlone and ar-curcumene. On the other side, Awasthi and Dixit [13] isolated the essential oil from the fresh turmeric rhizomes. The major components were ar-turmerone,  $\beta$ -turmerone and (Z)- $\beta$ -ocimene.

Regarding the composition of the essential oil obtained from the dried rhizomes grown in India, the major components were ar-turmerone, curlone, turmerone and ar-curcumene [16] while in the oil obtained from the dried turmeric rhizomes grown in China they were arturmerone, humulene oxide and  $\beta$ -selinene [18]. It could be concluded that the turmeric essential oil compositon, among many other factors, depends on geographic origin and part of the plant used.

Turmerone and ar-turmerone are, according to their chemical structure, ketonic sesquiterpenes of the bisabolane type responsible for turmeric aroma and smell [10]. Besides contributing to the turmeric organoleptic properties, turmerone posseses a wide range of pharmacological activities such as antioxidant, anti-inflammatory, anti-tumor, anti-proliferative and anti-depressant activity [26]. Ar-turmerone exhibits the mosquito repellent activity and it is an effective drug for the respiratory diseases treatment [27]. Curlone is used against hepatitis [27],

and allylbenzene eugenol is well known for its antibiotic properties [14]. Curcumin, on the other side, belongs to the group of curcuminoids i.e. diphenylheptanoids (low molecular weight polyphenols) and, as already mentioned, it is the most active and the most studied turmeric component. It acts as an antioxidant due to the capability to scavenge free radicals such as superoxide anions and hydroxyl radicals which are well known as potent initiators of lipid peroxidation. Among many other beneficial effects, curcumin exhibits the anti-inflammatory, anti-protozoal, anti-HIV and anti-tumor activity [12]. All these effects are due to the presence of hydroxyl and phenol groups, as well as the β-dicarbonylic system with conjugated double bonds (dienes). The mentioned system provides lipophylicity thus enabling curcumins' better skin and blood-brain barrier penetration. Curcumin is also well known as food colouring agent, with E number E100 [11,12].

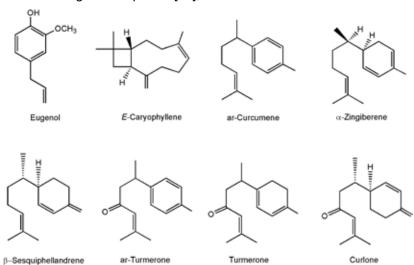


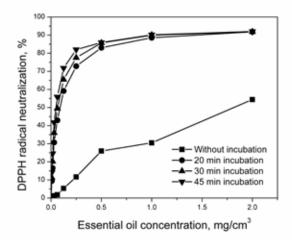
Figure 1. Structures of the identified components in turmeric essential oil

## Antioxidant activity

The antioxidant activity of the obtained essential oil was studied using DPPH assay. DPPH radical absorbs at 517 nm (violet in color). When exposing the radical to free radical scavengers its absorbance significantly decreases because of the hydrogen atom transfer from the antioxidant to the radical. Thereby, the absorbance decrease indicates on the antioxidant potential of the studied sample.

DPPH radical neutralization capacity depends on the applied oil concentration, as well as on the incubation time – it increases with the concentration of the not-incubated samples, while in the case of incubated samples this increase was up to the oil concentration of 0.5 mg/cm³ (the oil concentration added to the reaction mixture) and slightly increased later. The oil showed the best antioxidant activity (with the degree of DPPH radical neutralization of 92%) after 45 minutes of incubation. The oil incubated for 20 and 30 minutes showed a slightly lower degree of the DPPH radical neutralization in the same concentration (91.8%, for the both oils). On the other

side, the oil which was not incubated showed a considerably lower value, 54.4% (Figure 2.).



**Figure 2.** Antioxidant activity of the turmeric rhizome essential oil obtained using Clevenger hydrodistillation

Oil concentrations needed for the neutralization of 50% of the initial DPPH radical concentration (EC50 value) were 1.784 mg/cm³ (without incubation); 0.098 mg/cm³ (after 20 minutes); 0.072 mg/cm³ (after 30 minutes) and 0.045 mg/cm³ (after 45 minutes incubation with radical). The synthetic antioxidant BHT showed EC50 value of 0.021 mg/cm³ after 20 minutes incubation with DPPH

radical. Since BHT is one of the most used antioxidant but with deleterious effects on the human organism [28], based on the obtained results it could be concluded that the turmeric essential oil obtained in this study may be considered as a safer alternative to the synthetic antioxidants with potential application in pharmaceutical and food products.

Table 2. Antimicrobial activity of the essential oil from turmeric rhizome

	Facestial ail	Antibiotics						
Microorganism	Essential oil	Ampici <b>l</b> in	Bactrim	Cefalexin	Nystatin			
-	Inhibition zone, mm							
Candida albicans	31	n.t.	n.t.	n.t.	17			
Proteus vulgaris	15	13.2	22.9	n.t.	n.t.			
Bacillus cereus	17	n.t.	n.t.	n.t.	n.t.			
Bacillus subtilis	24	n.t.	n.t.	48	n.t.			
Klebsiella pneumoniae	18	n.t.	n.t.	13	n.t.			
Staphylococcus aureus	n.a.	36.7	42.1	26	n.t.			
Escherichia coli	n.a.	n.a.	15.0	26	n.t.			
Listeria monocitogenes	n.a.	n.t.	n.t.	34	n.t.			

n.a. – not active; n.t. – not treated; disk diameter 12.7 mm; oil content 60 μl (1:10 v/v in DMSO).

#### Antimicrobial activity

Turmeric essential oil showed activity upon the majority of the studied microorganisms. The strongest (antifungal) activity was observed on C. albicans. Additionally, the inhibition was almost two times stronger comparing with antifungal medication Nystatin - 31 mm vs. 17 mm, respectively (Table 2.). Candida albicans is a dimorphic fungus that makes normal human microbiota microflora in gastrointestinal and genitourinary tract. It is an opportunistic pathogen that attacks immunocompromised patients (for example after cancer chemotherapy or HIV infections), so candidiasis are very frequent infections. Considering that C. albicans is eukaryotic cell, it shares many common biologic properties with humans, so commercial antifungal agents used today cause harmful effects and it is necessary to develop new, more effective natural antifungal agents [29-31]. Considering that C. albicans is very persistent microorganism that causes urinary infection so obtained essential oil could be used not only in the preventive but also for therapeutic proposes.

*P. vulgaris* and *K. pneumoniae* belong to Gram-negative bacteria that cause urinary infections [32,33]. The obtained turmeric essential oil showed slightly better (18 mm) activity against *K. pneumoniae* than cephalosporin antibiotic Cefalexin (13 mm inhibition zone) (Table 2.).

Among Gram-positive bacteria (*B. cereus* and *B. sub-tilis*) *B. subtilis* was more sensitive. *Bacillus* species are responsible for numerous infections in humans due to the consumption of food products rich in starch and proteins, such as for example rice, meat and meat products. These bacteria also caused very serious diseases including meningitis, gangrene and eye infections [34].

The obtained essential oil showed no activity on *S. aureus*, *E. coli* and *L. monocitogenes*. Norajit and colleagues [35] extracted the essential oil by hydrodistillation from the fresh turmeric rhizome grown in Thailand. It showed the activity against *S. aureus*, but not against *E. coli*.

Regarding the inactivity of the obtained essential oil against *L. monocitogenes* our results are in agreement with those obtained by Antunes and coworkers [36]. In order to increase the activity of the essential oil against these bacteria they added an antioxidant - ascorbic acid and observed their synergistic action. It should be noted that antimicrobial activity of essential oils depends, among other, on the concentration and the chosen extraction method [37].

Generally, essential oils exhibit a stronger antimicrobial activity on Gram-positive bacteria (comparing to Gram-negative bacteria) because of the presence of lipoteichoic acids, i.e. their lipophilic ends in the (less complex, single-layer) membranes that facilitate the penetration of hydrophobic compounds present in essential oils. These compounds impart crucial processes in the cell due to the increased membrane permeability, consequently inducing leakage of ions and important cell contents, finally leading to the cell death. On the other side, Gram-negative bacteria are resistant thanks to the presence of extrinsic membrane proteins or cell wall lipopolysaccharides, with the ability of limitation of the diffusion rate of hydrophobic compounds through the (more complex, double layered - lipopolysaccharide) cell membrane [2,3].

The antimicrobial activity of the essential oil from turmeric rhizome is probably the consequence of common effects of all components present in the oil. However, it is difficult to compare the data with literature because the variables that influence the results are the essential oil composition and antimicrobial test method used. Moreover, the standard criteria for the evaluation of the plant essential oil activity are missing and therefore the results obtained by different authors are widely different [38].

#### Conclusion -

The essential oil was obtained from the turmeric commercial spice available in the local Serbian market. Eight compounds were identified and quantified by GC-MS and GC-FID analysis: eugenol, (E)-cariophyllen, ar-curcumen, α-zingiberene, β-sesquiphelandrene, ar-turmerone, turmerone and curlone. The composition of the obtained essential oil is very similar to that obtained from the turmeric rhizome grown in India. The major compound in the oil was turmerone (26%). Turmeric essential oil showed a high degree of DPPH radical neutralization. The concentration of the essential oil needed for neutralization of 50% of the initial DPPH radical concentration (EC50 value) was 0.045 mg/cm3 (after 45 minutes incubation with radical). A degree of DPPH radical neutralization of 92% was achieved by 2 mg/cm3 essential oil concentration. The obtained oil showed the best antimicrobial (antifungal) activity on C. albicans. Turmeric use is very rare in the Serbian traditional cuisine. Having in mind the obtained results, use of turmeric is not just reasonable but it should be even favored in the nutrition, as well as in food and pharmaceutical industries processing.

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Izvod -

# HEMIJSKI SASTAV, ANTIOKSIDATIVNA I ANTIMIKROBNA AKTIVNOST ETARSKOG ULJA KURKUME (*Curcuma longa* L.)

Jelena S. Stanojević, Ljiljana P. Stanojević, Dragan J. Cvetković, Bojana R. Danilović

Tehnološki fakultet, Leskovac, Srbija

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Ovaj rad se bavi ispitivanjem etarskog ulja dobijenog iz samlevenog rizoma kurkume (Rhizoma Curcumae) (Turska) Clevenger hidrodestilacijom pri hidromodulu 1:5 m/V u toku 180 minuta. Kvalitativni i kvantitativni sastav ulja je određen GC-MS i GC-FID spektrometrijom. Antioksidativna aktivnost dobijenog ulja je određena DPPH testom odmah nakon dodavanja DPPH radikala i nakon 20 min. 30 min i 45 min inkubacije sa radikalom. Antimikrobna aktivnost je određena diskdifuzionom metodom. Prinos etarskog ulja iznosi 0,3 cm<sup>3</sup>/100 g biljnog materijala. U ulju je identifikovano osam komponenti. Najzastupljenije komponente su ar-turmeron (22,7%), turmeron (26%) i kurlon (16,8%). Najbolju antioksidativnu aktivnost je pokazalo ulje inkubirano 45 minuta sa DPPH radikalom. EC50 vrednosti etarskog ulja iznose 1,784 mg/cm³ (bez inkubacije), 0,098 mg/cm³ (nakon 20 minuta), 0,072 mg/cm<sup>3</sup> (nakon 30 minuta) i 0,045 mg/cm<sup>3</sup> (nakon 45 minuta inkubacije sa radikalom). Dobijeno etarsko ulje je pokazalo najbolju antimikrobnu (antifungalnu) aktivnost na Candidu albicans. Dobijeni rezultati pokazuju da je etarsko ulje kurkume izuzetno jak antioksidans i antimikrobni (antifungalni) agens sa potencijalnom primenom u prehrambenoj i farmaceutskoj industriji kao bezbednija alternativa sintetskim antioksidansima i antimikrobnim agensima.

**Ključne reči:** *Curcuma longa* L., Clevenger hidrodestilacija, antioksidativna aktivnost, antimikrobna aktivnost, GC-MS